

(19)



Europäisches Patentamt  
European Patent Office  
Office européen des brevets



(11) Publication number:

**0 416 564 A2**

(12)

**EUROPEAN PATENT APPLICATION**

(21) Application number: 90117020.9

(51) Int. Cl.<sup>5</sup>: **C07C 49/84, C07C 49/76,  
C07C 49/82, C07C 233/11,  
A61K 7/42, A61K 7/00**

(22) Date of filing: 04.09.90

(30) Priority: 05.09.89 JP 229708/89  
22.01.90 JP 12196/90

(43) Date of publication of application:  
13.03.91 Bulletin 91/11

(84) Designated Contracting States:  
DE ES FR GB

(71) Applicant: **KAO CORPORATION**  
1-14-10, Nihonbashikayaba-cho  
Chuo-ku Tokyo(JP)

(72) Inventor: Yamada, Shinji  
4594, Ichihana, Ichikaimachi  
Haga-gun, Tochigi(JP)  
Inventor: Kawamata, Akira  
8-3, 1-chome, Daishinden  
Utsunomiya-shi, Tochigi(JP)  
Inventor: Imokawa, Genji

1022-89, Himuro-cho  
Utsunomiya-shi, Tochigi(JP)  
Inventor: Masuda, Shinichi  
6-100, Dojojuku-machi  
Utsunomiya-shi, Tochigi(JP)  
Inventor: Yamaguchi, Masakazu  
4594, Ichihana, Ichikai-machi  
Haga-gun, Tochigi(JP)  
Inventor: Niinaka, Koichi  
4594, Ichihana, Ichikai-machi  
Haga-gun, Tochigi(JP)  
Inventor: Joukura, Hiroko  
561-10, Yamamoto-cho  
Utsunomiya-shi, Tochigi(JP)

(74) Representative: Wächtershäuser, Günter, Dr.  
Tal 29  
W-8000 München 2(DE)

(54) Novel aroyl ketone derivative, UV ray absorber comprising the same, and cosmetic composition containing the same.

(57) A novel aroyl ketone derivative represented by formula (I) is disclosed.

$\text{Ar}(\text{COCH}_2\text{COR}^1)_n$  (I)

wherein R<sup>1</sup> represents a saturated or unsaturated hydrocarbon group having 1-24 carbon atoms, or a linear or branched alkoxyalkyl or alkenyloxyalkyl group having 1-24 carbon atoms, dialkylamino group having 1-24 carbon atoms, Ar represents a benzene or naphthalene ring which may be substituted with one or more hydroxyl groups, alkoxy groups having 1-24 carbon atoms, alkenyloxy groups having 1-24 carbon atoms, polyoxyleneoxy groups, or  $\alpha$ -methylenedioxy groups, and n denotes an integer of 1-4. An UV-ray absorber and cosmetic composition comprising the aroyl ketone derivative of formula (I) are also disclosed. Aroyl ketone derivative (I) of the present invention can provide an excellent UV-ray absorbing effect and has outstanding stability against light, and the UV-ray absorber or cosmetic composition containing the aroyl ketone derivative (I) can exhibit an excellent sunscreensing effect.

EP 0 416 564 A2

# NOVEL AROYL KETONE DERIVATIVE, UV RAY ABSORBER COMPRISING THE SAME, AND COSMETIC COMPOSITION CONTAINING THE SAME

## BACKGROUND OF THE INVENTION

### 5 Field of the Invention:

This invention relates to a novel aroyl ketone derivative possessing UV-ray absorbing ability and outstanding stability against light, and to a UV-ray absorber and a cosmetic composition both containing such an aroyl ketone derivative and exhibiting a superior effect of protecting the skin from sunburn.

10

### Description of the Background Art:

Ultraviolet rays are known to cause various changes to the skin. In dermatology, UV-rays are classified  
15 into (a) long wavelength UV-rays of 400 - 320 nm, (b) medium wavelength UV-rays of 320 - 290 nm, and (c) short wavelength UV-rays of less than 290 nm. These are called UV-A, UV-B, and UV-C, respectively. The UV-rays contained in the sunlight are UV-A and UV-B, whereas UV-C is absorbed in the ozone layer and therefore scarcely reaches the ground.

The UV-B, when irradiated onto the skin in an amount exceeding a certain level, causes red spots or  
20 blisters on the skin. In addition, the UV-B tends to promote the formation of melanin, resulting in the melanoses or other adverse changes of the skin. On the other hand, UV-A induces melanism on the skin immediately after irradiation (instantaneous melanism). In addition, UV-A energy reaches the corium and brings about chronic changes in the blood vessel walls and the elastic fibers in connective tissues. Such behaviors of both UV-A and UV-B are considered to accelerate aging of the skin and to be a cause of  
25 wrinkles, spots or freckles.

As such effects of UV-rays on human skin have been clarified, various chemical compounds capable of absorbing UV-A and/or UV-B have been developed. Such UV-ray absorbers are desired to satisfy the following requirements (1) through (5):

- (1) absorbing UV-A or UV-B light to the maximum.
- 30 (2) having good stability against heat and light.
- (3) being free from toxicity, irritation, and any other harmful actions to the skin.
- (4) having a long-lasting effect.
- (5) having an excellent compatibility with cosmetic base materials.

Conventionally, for example, dibenzoylmethane derivatives are used as a UV-A absorber; and deriva-  
35 tives of cinnamate, benzophenone, p-aminobenzoic acid, salicylic acid, and the like are used as a UV-B absorber.

However, these conventional UV-ray absorbers can not necessarily satisfy the above requirements. Specifically, these UV-ray absorbers are known to be inadequate in the stability against light and to decompose or react by UV-ray irradiation (*Int. J. Cosmetic Science*, 10, 53, 1988). Such decomposition of  
40 UV-ray absorbers substantially decreases the lasting effect of compositions. Adverse effects to the skin resulting from decomposed substances themselves or compounds produced by the reaction of the decomposed substances and other components can not be neglected (*Fragrance Journal*, 84, 34, 1987).

Development of improved UV-ray absorbers which satisfy the above characteristics, especially UV-ray absorbers having excellent stability against light, has therefore been desired.

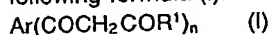
45 The inventors have synthesized many diketone derivatives and examined their characteristics, and, as a result, found that an aroyl ketone derivative represented by formula (I) shown below had an outstanding UV-A and UV-B absorbing ability and excellent stability against light, and further found that a cosmetic composition comprising the aroyl ketone derivative could efficiently protect the skin from being sunburnt.

50

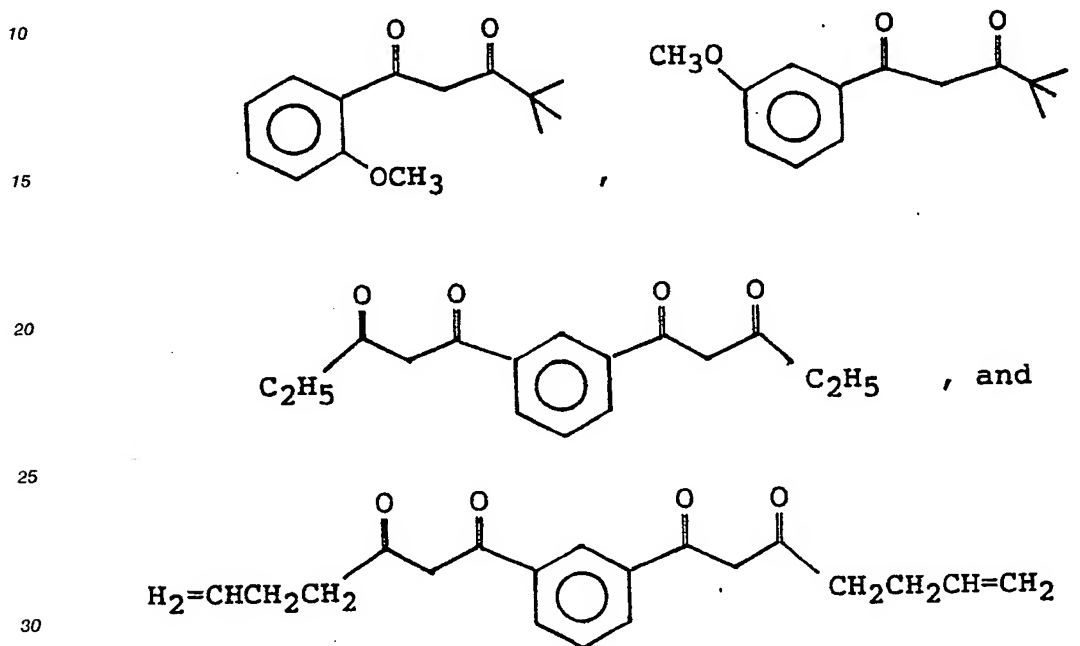
## SUMMARY OF THE INVENTION

Accordingly, an object of this invention is to provide a novel aroyl ketone derivative represented by the

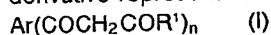
following formula (I):



wherein  $\text{R}^1$  represents a saturated or unsaturated hydrocarbon group having 1-24 carbon atoms, or a linear or branched alkoxyalkyl or alkenyloxyalkyl group having 1-24 carbon atoms, dialkylamino group having 1-24 carbon atoms, Ar represents a benzene or naphthalene ring which may be substituted with one or more hydroxyl groups, alkoxy groups having 1-24 carbon atoms, alkenyloxy groups having 1-24 carbon atoms, or  $\alpha$ -methylenedioxy groups, and n denotes an integer of 1-4; excluding the compounds represented by the following formulae:

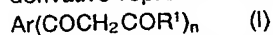


Another object of the present invention is to provide a UV-ray absorber comprising an aroyl ketone derivative represented by the following formula (I):



wherein  $\text{R}^1$  represents a saturated or unsaturated hydrocarbon group having 1-24 carbon atoms, or a linear or branched alkoxyalkyl or alkenyloxyalkyl group having 1-24 carbon atoms, dialkylamino group having 1-24 carbon atoms, Ar represents a benzene or naphthalene ring which may be substituted with one or more hydroxyl groups, alkoxy groups having 1-24 carbon atoms, alkenyloxy groups having 1-24 carbon atoms, or  $\alpha$ -methylenedioxy groups, and n denotes an integer of 1-4.

Still another object of this invention is to provide a cosmetic composition comprising an aroyl ketone derivative represented by the following formula (I):



wherein  $\text{R}^1$  represents a saturated or unsaturated hydrocarbon group having 1-24 carbon atoms, or a linear or branched alkoxyalkyl or alkenyloxyalkyl group having 1-24 carbon atoms, Ar represents a benzene or naphthalene ring which may be substituted with one or more hydroxyl groups, alkoxy groups having 1-24 carbon atoms, alkenyloxy groups having 1-24 carbon atoms, or  $\alpha$ -methylenedioxy groups, and n denotes an integer of 1-4.

Other objects, features and advantages of the invention will hereinafter become more readily apparent from the following description.

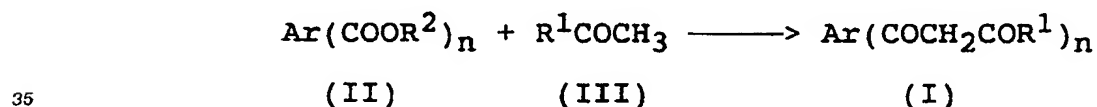
### DETAILED DESCRIPTION OF THE INVENTION AND PREFERRED EMBODIMENTS

Enumerated as examples of  $\text{R}^1$  having 1-24 carbon atoms, saturated or unsaturated, are methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, t-butyl, butenyl, n-pentyl, n-hexyl, n-heptyl, heptenyl, n-octyl, octenyl, n-nonyl, nonenyl, n-decyl, decenyl, n-undecyl, undecenyl, n-dodecyl, dodecenyl, n-tridecyl, tridecenyl, n-

tetradecyl, tetradecenyl, n-pentadecyl, pentadecenyl, isopentadecyl, n-hexadecyl, hexadecenyl, isohexadecyl, n-heptadecyl, heptadecenyl, isoheptadecyl, n-octadecyl, isooctadecyl, cyclohexyl, and adamantyl groups and the like. Enumerated as examples of hydroxyalkyl groups having 1-18 carbon atoms are hydroxymethyl, hydroxyethyl, hydroxypropyl, hydroxybutyl, hydroxypentyl, hydroxyhexyl, hydroxyheptyl, hydroxyoctyl, hydroxynonyl, hydroxydecyl, hydroxyundecyl, hydroxydodecyl, hydroxytetradecyl, hydroxyhexadecyl, hydroxyoctadecyl groups and the like. Enumerated as examples of linear or branched alkoxyalkyl groups having 1-24 carbon atoms are methoxymethyl, ethoxymethyl, propoxymethyl, isopropoxymethyl, butoxymethyl, pentyloxymethyl, hexyloxymethyl, octyloxymethyl, decyloxymethyl, undecyloxymethyl, tetradecyloxymethyl, hexadecyloxymethyl, octadecyloxymethyl, methoxyethyl, ethoxyethyl, allyloxyethyl, propoxyethyl, methoxypropyl, ethoxypropyl, allyloxypropyl, propoxypropyl, methoxybutyl, ethoxybutyl, allyloxybutyl, propoxybutyl, methoxypentyl, ethoxypentyl, allyloxypentyl, propoxypentyl, methoxydodecyl, ethoxydodecyl, allyloxydodecyl, propoxydodecyl, methoxyhexadecyl, ethoxyhexadecyl, allyloxyhexadecyl, propoxyhexadecyl, methoxyhexadecyl, ethoxyoctadecyl, allyloxyoctadecyl, propoxyoctadecyl groups, and the like. Given as examples of alkenyloxyalkyl groups having 1-24 carbon atoms are allyloxymethyl, butenyloxymethyl, hexenyloxymethyl, undecenylloxymethyl, octadecenylloxymethyl groups, and the like. Enumerated as examples of dialkylamino groups having 1-24 carbon atoms are dimethylamino, ethylmethylamino, diethylamino, methylpropylamino, methylisopropylamino, methylbutylamino, methyl-*t*-butylamino, diisopropylamino, dipropylamino, ethylbutylamino, methylhexylamino, di-*sec*-butylamino, diisobutylamino, dipentylamino, dihexylamino, bis 2-ethylhexylamino, dioctylamino, methyloctadecylamino, pyrrolidyl, piperidyl, morpholinyl groups, and the like. Preferable examples of linear or branched alkoxy groups having 1-24 carbon atoms, by which the benzene or naphthalene ring (Ar) may be substituted, are those having 1-8 carbon atoms, for example, methoxy, ethoxy, *n*-propoxy, isopropoxy, *n*-butoxy, and *tert*-butoxy groups, and the like. Preferable examples of alkenyloxy groups having 1-24 carbon atoms, by which the benzene or naphthalene ring (Ar) may be substituted, are those having 1-8 carbon atoms, for example, allyloxy, butenyloxy, pentenyloxy and hexenyloxy groups, and the like. *n* is an integer of 1-4, but its preferable range is 1-3.

Aroyl ketone derivative (I) of the present invention can be produced by one of the following processes (a), (b) or (C), according to known methods, for example, reported in *J. Am. Chem. Soc.*, **80**, 4891, (1958); *J. Chromatogr.*, **312**, 109, (1984); and *J. Polym. Sci. Polym. Chem. Ed.*, **20**, 3079, (1982).

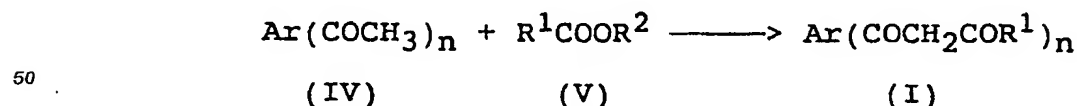
Process (a):



wherein  $\text{R}^2$  represents a methyl group, ethyl group, *n*-propyl group, or a *n*-butyl group and  $\text{R}^1$  and *n* have the same meanings as defined above.

As is clear from the above reaction scheme, compound (I) of the present invention can be prepared by condensation of an ester of formula (II) and a ketone of formula (III). In process (a), it is desirable that the reaction be carried out at 20-150°C for tens of minutes to 10 hours in a solvent such as anhydrous tetrahydrofuran, toluene, xylene, or the like using a base as a catalyst. The base used in this process is a metallic hydride, e.g. sodium hydride; an alkyl metal, e.g. butyl lithium; an amine, e.g. triethylamine; a metal amide, e.g. sodium amide; a metal alkoxyde, e.g. sodium methoxide; and the like.

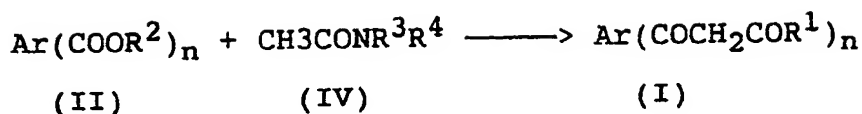
Process (b):



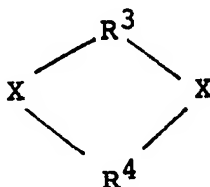
wherein  $\text{R}^1$ ,  $\text{R}^2$ , and *n* have the same meanings as defined above.

As is clear from the above reaction scheme, compound (I) of the present invention can be prepared by condensation of an acetylbenzene derivative or an acetylnaphthalene derivative of formula (IV) and an ester of formula (V). The reaction of process (b) can be carried out under the same conditions as in process (a).

Process (c):



wherein R<sup>1</sup>, R<sup>2</sup>, and n have the same meanings as defined above. R<sup>3</sup> and R<sup>4</sup> each represents a hydrocarbon group having 1-18 carbon atoms, which may form a ring such as,



wherein X is oxygen.

As is clear from the above reaction scheme, compound (I) of the present invention can be prepared by condensation of an ester of formula (II) and an acetoamide of formula (IV). The reaction of process (c) can be carried out under the same conditions as in process (a).

The UV-ray absorbing agent of the present invention may consist of one or more aroyl ketone derivatives of formula (I) thus obtained. However, it is more desirable to use the aroyl ketone derivatives of the present invention mixed with a carrier. Any kind of carriers which are inert to the aroyl ketone derivative (I) of the present invention and in the form either of solid, liquid, emulsion, foam, gel, or the like can be used. Typical examples of these carriers are water; alcohols; oil and fats, e.g. hydrocarbon oils, fatty acid esters, long-chain alcohols, silicone; micro powders of starch, talc, and the like; low boiling point hydrocarbons or halogenated hydrocarbons used as a conventional aerosol propellant; and the like. In addition to the above components, any conventional components may be incorporated in the UV-ray absorber of the present invention to the extent that the UV-ray absorbing effect of the aroyl ketone derivative of the present invention will not be impaired. These conventional components include, for example, antiseptics, perfumes, pigments, surfactants, and the like.

The cosmetic compositions containing the aroyl ketone derivative (I) of the present invention (hereinafter designated as "cosmetic composition of the present invention") can be produced by incorporating one or more aroyl ketone derivatives (I) having compatibility with the base components of the cosmetic composition according to a conventional method. The cosmetic composition can be prepared into various forms such as creams, liquids, oils, sprays, sticks, emulsions, foundations, ointments, and the like.

The selective use of aroyl ketone derivatives (I) which are compatible with the cosmetic bases ensures preparation of cosmetic compositions having UV-ray absorbing effect of various types, for example, fundamental cosmetics such as cosmetic oils with oil bases, oil creams and milky lotions with a large quantity of oil, weakly oily creams and lotions, water-base lotions; and make-up cosmetics such as oil-base foundations, lipsticks. Examples of these base components or solvents useful for the cosmetic compositions of the present invention are hydrocarbons such as solid or liquid paraffins, crystal oils, ceresin, ozokerite, montan wax, and the like; oils, fats, or waxes of plant or animal origin such as olive oil, earth wax, carnauba wax, lanolin, spermaceti, and the like; fatty acids and esters thereof such as stearic acid, palmitic acid, oleic acid, glycerol monostearate, glycerol distearate, glycerol monooleate, isopropyl myristate, isopropyl stearate, butyl stearate, and the like; alcohols such as ethyl alcohol, isopropyl alcohol, cetyl alcohol, stearyl alcohol, palmityl alcohol, hexyldodecyl alcohol, and the like. In addition, polyhydric alcohols having a moisturizing effect such as glycol, glycerol, sorbitol, and the like may also be used.

The amount of aroyl ketone derivative (I) to be incorporated in the UV-ray absorbing agent or cosmetic composition of the present invention varies depending on their types or forms of products. There are no specific limitations as to the amount to be added in inasmuch as such an addition brings about the effect of the present invention. A preferable amount is 0.1-20% by weight, and particularly 0.5-10% by weight, of the crude product.

The UV-ray absorbing agent or cosmetic composition of the present invention can be prepared by simply incorporating the aroyl ketone derivative (I) as a UV-ray absorber. However, it is more preferable to provide a sunscreen cosmetics for general use by incorporating conventional UV-A or UV-B absorbers together with the aroyl ketone derivative (I) of this invention. Examples of these conventional UV-B

absorbers include, p-methylbenzylidene-D,L-camphor or its sodium sulfonate, sodium 2-phenylbenzimidazole-5-sulfonate, sodium 3,4-dimethylphenylglyoxylate, 4-phenylbenzophenone, isooctyl 4-phenylbenzophenone-2'-carboxylate, p-methoxycinnamate, 2-phenyl-5-methylbenzoxazol, p-dimethylaminobenzoate, and the like. Examples of the UV-A absorbers are 4-methoxy-2'-carboxydibenzoylmethane, 4-methoxy-4'-t-butylidibenzoylmethane, 4-isopropylidibenzoylmethane, 2-hydroxy-4-methoxybenzophenone, dibenzylidenecamphor, and the like.

Into the cosmetic compositions of the present invention, various additives other than the above-mentioned components can be incorporated. Suitable additives are, for example, W/O and O/W emulsifiers which are available in the market; and viscosity increasing agents such as methyl-, ethyl-, or carboxymethyl cellulose, polyacrylic acid, tragacanth, agar, gelatin, and the like. As required, perfumes, antiseptics, humectants, emulsion stabilizers, medical components, and/or coloring agents which are physiologically acceptable, may also be formulated into the cosmetic composition of this invention.

Other features of the invention will become apparent in the course of the following description of the exemplary embodiments which are given for illustration of the invention and are not intended to be limiting thereof.

### EXAMPLES

#### Example 1

#### Synthesis of 3,4-dimethoxy(4,4-dimethyl-3-oxopentanoyl)benzene:

In a 200 ml three-necked flask equipped with a mechanical stirrer, dropping funnel, reflux condenser, and a nitrogen-inlet tube, 2.45 gm (61 mmol) of 60% sodium hydride, 10 gm (51 mmol) of methyl 3,4-dimethoxybenzoate, and 100 ml of anhydrous tetrahydrofuran were mixed with stirring under nitrogen stream, and refluxed with heating while 6.1 gm (61 mmol) of pinacolone was added dropwise. The refluxing under heat was continued for 7 hours. After cooling the reaction mixture, 30 ml of 2N hydrochloric acid was added and the mixture was extracted twice with chloroform. The extract was dried over anhydrous sodium sulfate and the solvent was removed by evaporation to give a crude product. Hexane was added to the crude product and insoluble substances were filtered off. The filtrate was concentrated by evaporation, and recrystallization afforded 8.9 gm of the target compound as colorless needles (yield: 65%).

Melting Point: 52.3-53.3 °C

IR( $\nu_{\text{KBr}}$ ,  $\text{cm}^{-1}$ ): 1602, 1521, 1470, 1446, 1365, 1299, 1266, 1218, 1188, 1131, 885, 786, 729

$^1\text{H-NMR}$ ( $\text{CDCl}_3$ ,  $\delta$ ): 1.26(9H, s, t- $\text{C}_4\text{H}_9$ ), 3.95(3H, s,  $\text{OCH}_3$ ), 3.96(3H, s,  $\text{OCH}_3$ ), 6.24(1H, s), 6.90(1H, d,  $J=8.4\text{Hz}$ ), 7.49(1H, s) 7.51(1H, d,  $J=8.4\text{Hz}$ )

Elemental analysis	
Calculated (%)	C: 68.16, H: 7.63
Found (%)	C: 68.23, H: 7.60

#### Example 2

#### Synthesis of 3,4,5-trimethoxy(4,4-dimethyl-3-oxopentanoyl)benzene:

In a 200 ml three-necked flask equipped with a mechanical stirrer, dropping funnel, reflux condenser, and a nitrogen-inlet tube, 3.0 gm (75 mmol) of 60% sodium hydride, 10 gm (44.3 mmol) of methyl 3,4,5-trimethoxybenzoate, and 80 ml of anhydrous tetrahydrofuran were mixed with stirring under nitrogen stream, and refluxed with heating while 5.3 gm (53.2 mmol) of pinacolone was added dropwise. Refluxing under heat was continued for 5 hours. After cooling the reaction mixture, 45 ml of 2N hydrochloric acid was

added, the mixture was extracted twice with chloroform. The extract was dried over anhydrous sodium sulfate and the solvent was removed by evaporation to give a crude product. Hexane was added to the crude product and insoluble substances were filtered off. The filtrate was concentrated by evaporation, and recrystallization afforded 9.6 gm of the target compound as colorless needles (yield: 74%).

5 Melting Point: 67.3-68.4 °C

IR( $\nu_{\text{KBr}}$ ,  $\text{cm}^{-1}$ ): 2968, 1593, 1560, 1509, 1473, 1434, 1335, 1233, 1218, 1176, 1131, 993, 801

$^1\text{H-NMR}(\text{CDCl}_3, \delta)$ : 1.26(9H, s,  $\text{t-C}_4\text{H}_9$ ), 3.91(3H, s,  $\text{OCH}_3$ ), 3.93(6H, s,  $\text{OCH}_3$ ), 6.21(1H, s), 7.13(2H, s)

Elemental analysis	
Calculated (%)	C: 65.29, H: 7.53
Found (%)	C: 65.38, H: 7.51

### Example 3

#### 20 Synthesis of 1,4-bis(4,4-dimethyl-3-oxopentanoyl)benzene:

In a 2 l four-necked flask equipped with a mechanical stirrer, dropping funnel, thermometer, reflux condenser, and a nitrogen-inlet tube, 32 gm (0.8 mol) of 60% sodium hydride, 79 gm (0.79 mol) of pinacolone, and 700 ml of anhydrous tetrahydrofuran were mixed with stirring under nitrogen stream. To this solution, 70 gm (0.36 mol) of dimethylterephthalate in 300 ml of tetrahydrofuran was added dropwise over one hour. After the addition, the reaction mixture was refluxed with heating for 6 hours, and was cooled to room temperature. After addition of 180 ml of 2N aqueous hydrochloric acid, the reaction mixture was extracted with chloroform. The extract was dried over anhydrous sodium sulfate and the solvent was evaporated to give a crude product. The crude product was recrystallized using hexane to obtain 78 gm of the target compound as colorless scales (yield: 66%).

Melting Point: 125.5-126.5 °C

IR( $\nu_{\text{KBr}}$ ,  $\text{cm}^{-1}$ ): 2974, 2872, 1584, 1563, 1485, 1368, 1290, 1140, 840, 792, 741

$^1\text{H-NMR}(\text{CDCl}_3, \delta)$ : 1.27(18H, s), 6.34(2H, s), 7.95(4H, s)

Elemental analysis	
Calculated (%)	C: 72.70, H: 7.93
Found (%)4	C: 72.61, H: 7.96

### Example 4

#### Synthesis of 1,3-bis(4,4-dimethyl-3-oxopentanoyl)benzene:

In a 100 ml three-necked flask equipped with a mechanical stirrer, dropping funnel, reflux condenser, and a nitrogen-inlet tube, 1.5 gm (38 mmol) of 60% sodium hydride, 3.6 gm (36 mmol) of pinacolone, 3.0 gm (15.5 mmol) of dimethyl isophthalate, and 30 ml of anhydrous tetrahydrofuran were mixed with stirring under nitrogen stream and refluxed with heating for 6 hours. After cooling the reaction mixture, 10 ml of 2N aqueous hydrochloric acid was added the product was extracted with chloroform. The extract was dried over anhydrous sodium sulfate and the solvent was evaporated to give a crude product. Recrystallization using hexane afforded 3.1 gm of the target compound as colorless crystals (yield: 61%).

Melting Point: 106.0-107.5 °C

IR( $\nu_{\text{KBr}}$ ,  $\text{cm}^{-1}$ ): 3124, 2974, 2872, 1611, 1563, 1482, 1431, 1290, 1227, 1134, 1095, 879, 804, 705

$^1\text{H-NMR}(\text{CDCl}_3, \delta)$ : 1.27(18H, br.s), 6.34(2H, s), 7.55(1H, br.t,  $J=7.8\text{Hz}$ ), 8.00 and 8.04(2H, dd,  $J=7.8\text{Hz}$ ),

J = 1.5Hz), 8.38(1H, br.s)

Elemental analysis	
Calculated (%)	C: 72.70, H: 7.93
Found (%)	C: 72.58, H: 7.95

#### Example 5

##### Synthesis of 4-methoxy-1,3-bis(4,4-dimethyl-3-oxopentanoyl)benzene:

In a 100 ml three-necked flask equipped with a mechanical stirrer, dropping funnel, reflux condenser, and a nitrogen-inlet tube, 1.5 gm (38 mmol) of 60% sodium hydride, 3.6 gm (36 mmol) of pinacolone, 3.0 gm (13.4 mmol) of dimethyl 4-methoxyisophthalate, and 30 ml of anhydrous tetrahydrofuran were mixed with stirring under nitrogen stream and refluxed with heating for 5 hours. After cooling the reaction mixture, 10 ml of 2N aqueous hydrochloric acid was added, and the product was extracted with chloroform. The extract was dried over anhydrous sodium sulfate and the solvent was removed by evaporation. The crude product was recrystallized using hexane to obtain 3.2 gm of the target compound as colorless crystals (yield: 66%).

Melting Point: 69.4-70.8 °C

IR( $\nu_{\text{KBr}}$ ,  $\text{cm}^{-1}$ ): 2968, 1620, 1584, 1506, 1467, 1368, 1275, 1263, 1182, 1131, 1071, 1011, 795

$^1\text{H-NMR}$ ( $\text{CDCl}_3$ ,  $\delta$ ): 1.24(9H, s), 1.25(9H, s), 3.99(3H, s), 6.29(1H, s), 6.52(1H, s), 7.04(1H, d,  $J = 8.8\text{Hz}$ ), 8.04(1H, dd,  $J = 2.3\text{Hz}$ ,  $J = 8.8\text{Hz}$ ), 8.35(1H, d,  $J = 2.3\text{Hz}$ )

Elemental analysis	
Calculated (%)	C: 69.98, H: 7.83
Found (%)	C: 69.92, H: 7.85

#### Example 6

##### Synthesis of 2,6-bis(4,4-dimethyl-3-oxopentanoyl)-naphthalene:

To 2.0 gm of dimethyl 2,6-naphthalene dicarboxylate and 2.8 gm of pinacolone, which had been dissolved in 25 ml of anhydrous tetrahydrofuran, 1.1 gm of 60% sodium hydride was added. The mixture was reacted at 60 °C for 4 hours while stirring. The reaction mixture was poured into ice-cooled 2N hydrochloric acid. Organic substances were extracted using chloroform and, after evaporating the solvent, recrystallized in a mixed solvent of chloroform-methanol to obtain 2.1 gm of the target compound as pale yellow crystals.

Melting Point: 195-196 °C

IR( $\nu_{\text{KBr}}$ ,  $\text{cm}^{-1}$ ): 2968, 2936, 2872, 1618, 1580, 1512, 1480, 1430, 1182, 1154, 1128, 1070, 1018, 910, 812, 710, 474

$^1\text{H-NMR}$ ( $\text{CDCl}_3$ ,  $\delta$ ): 15.81(2H, s), 8.62(2H, s), 7.94-8.10(4H, m), 6.47(2H, s), 1.30(18H, s)

#### Example 7

##### Synthesis of 1,4-bis(4-methyl-3-oxopentanoyl)-benzene:



The procedures in Example 3 were repeated under substantially same conditions except that 68 gm (0.79 mol) of isopropylmethylketone was used instead of 79 gm (0.79 mol) of pinacolone, and afforded 76 gm of the target compound as pale yellow needles (Yield: 72%).

Melting Point: 97.0-97.5 °C

- 5 IR (KBr,  $\text{cm}^{-1}$ ): 2980, 2932, 1608, 1437, 1284, 1188, 1098, 939, 807  
 $^1\text{H-NMR}(\text{CDCl}_3, \delta)$ : 1.25(12H, d), 2.55-2.77(2H, m), 6.25(2H, s), 7.95(4H, s), 14.7(2H, bs)

#### Example 8

10

Synthesis of 1,4-bis(3-oxopentanoyl)benzene:

- 15 The procedures in Example 3 were repeated under substantially same conditions except that 57 gm (0.79 mol) of methylethylketone was used instead of 79 gm (0.79 mol) of pinacolone, and afforded 54 gm of the target compound as pale yellow crystals (Yield: 56%).

Melting Point: 122.5-123.5 °C

IR (KBr,  $\text{cm}^{-1}$ ): 2980, 2950, 1617, 1293, 1161, 1119, 1083, 813, 774  
 $^1\text{H-NMR}(\text{CDCl}_3, \delta)$ : 1.25(6H, t), 2.50(4H, q), 6.22(2H, s), 7.95(4H, s), 15.2(2H, bs),

20

#### Example 9

- 25 Synthesis of 1,4-bis(3-oxododecanoyl)benzene:

- In a 100 ml three-necked flask equipped with a mechanical stirrer, dropping funnel, reflux condenser, and a nitrogen-inlet tube, 1.1 gm (28 mmol) of 60% sodium hydride, 2.0 gm (12.3 mmol) of p-diacetylbenzene, 4.8 gm (25.7 mmol) of methylcaprylate, and 20 ml of anhydrous tetrahydrofuran were mixed with stirring under nitrogen stream and the refluxing under heat was continued for 5 hours. After cooling the reaction mixture, 15 ml of 2N aqueous hydrochloric acid was added, and the mixture was extracted with chloroform. The extract was dried over anhydrous sodium sulfate and the solvent was removed by evaporation to give a crude product. Hexane was added to the crude product and recrystallization afforded 3.6 gm of the target compound as pale yellow scales (yield: 63%).

- 35 Melting Point: 122.5-123.0 °C

IR (KBr,  $\text{cm}^{-1}$ ): 2920, 2854, 1617, 1473, 1293, 1155, 786  
 $^1\text{H-NMR}(\text{CDCl}_3, \delta)$ : 1.90(6H, t), 1.12-1.43(20H, m), 1.56(4H, bs), 1.60-1.80(4H, m), 2.47(4H, t), 6.20(2H, s), 7.95(4H, s), 14.6(2H, bs)

40

#### Example 10

Synthesis of 1,3,5-tris(4,4-dimethyl-3-oxopentanoyl)benzene:

45

- In a 100 ml three-necked flask equipped with a mechanical stirrer, dropping funnel, reflux condenser, and a nitrogen-inlet tube, 2.29 gm (57 mmol) of 60% sodium hydride, 5.24 gm (52 mmol) of pinacolone, 4.0 gm (15.8 mmol) of trimethyl-1,3,5-benzenetricarboxylate, and 40 ml of anhydrous tetrahydrofuran were mixed with stirring under nitrogen stream, and the refluxing under heat was continued for 7 hours. After cooling the reaction mixture, 30 ml of 2N aqueous hydrochloric acid was added, and the mixture was extracted with chloroform. The extract was dried over anhydrous sodium sulfate and the solvent was removed by evaporation to give a crude product. Acetone was added to the crude product, and recrystallization afforded 3.8 gm of the target compound as yellow crystals (yield: 52%).

- 50  $^1\text{H-NMR}(\text{CDCl}_3, \delta)$ : 1.30(27H, s), 6.41(3H, s), 8.49(3H, s), 15.6(3H, bs)

55

#### Example 11

## Synthesis of 1,4-bis(4-hydroxy-4-methyl-3-oxopentanoyl)benzene:

In a 200 ml three-necked flask equipped with a mechanical stirrer, dropping funnel, reflux condenser, and a nitrogen-inlet tube, 15 gm (147 mmol) of 3-methyl-3-hydroxy-2-butanone, 500 mg (2.0 mmol) of pyridinium p-toluenesulfonate and 100 ml of anhydrous methylenechloride were mixed with stirring under nitrogen stream, and 11.7 gm (162 mmol) of ethylvinylether was added dropwise at room temperature for 3 hours while stirring. After the reaction, the reaction mixture was washed with 5% aqueous sodium hydrogencarbonate, and then the mixture was extracted with chloroform. The extract was dried over anhydrous sodium sulfate and the solvent was removed by evaporation to give a crude product. Distillation of the crude product afforded 17.09 gm of 3-(1-ethoxyethoxy)-3-methyl-butane-2-one (b.p. 38-40/1 mmHg, yield: 66%).

In a 100 ml three-necked flask equipped with a mechanical stirrer, dropping funnel, reflux condenser, and a nitrogen-inlet tube, 9.9 gm (57 mmol) of above-mentioned 3-(1-ethoxyethoxy)-3-methyl-butane-2-one, 2.3 gm (57 mmol) of 60% sodium hydride, 5.0 gm (25 mmol) of dimethylterephthalate and 50 ml of anhydrous tetrahydrofuran were mixed with stirring under nitrogen stream, and the refluxing under heat was continued for 7 hours. After cooling the reaction mixture, 30 ml of 2N hydrochloric acid was added and the mixture was extracted with chloroform. The extract was dried over anhydrous sodium sulfate and the solvent was removed by evaporation. The reaction mixture was stirred at room temperature while adding 30 ml of methanol with ten drops of 2N hydrochloric acid. A mixed solvent of acetone-chloroform was added to the crude product, and recrystallization afforded 4.1 gm of the target compound as pale yellow crystals (yield: 48%).

Melting Point: 276.5-277.5 °C

IR( $\nu_{\text{KBr}}$ ,  $\text{cm}^{-1}$ ): 3112, 2980, 1690, 1583, 1425, 1356, 1173, 1056, 825

$^1\text{H-NMR}$ ( $\text{CDCl}_3$ ,  $\delta$ ): 1.52(12H, s), 6.09(2H, s), 7.95(4H, s), 14.8(2H, bs)

## Example 12

## 30 Synthesis of 1,4-bis(4-methyl-3-oxo-4-azapentanoyl)benzene:

In a 200 ml three-necked flask equipped with a mechanical stirrer, dropping funnel, reflux condenser, and a nitrogen-inlet tube, 5.0 gm (125 mmol) of 60% sodium hydride, 10.6 gm (122 mmol) of dimethylacetamide, 10 gm (51 mmol) of dimethylterephthalate and 100 ml of anhydrous tetrahydrofuran were mixed with stirring under nitrogen stream, and the refluxing under heat was continued for 4 hours. After cooling the reaction mixture, 65 ml of 2N aqueous hydrochloric acid was added and the mixture was extracted with chloroform. The extract was dried over anhydrous sodium sulfate and the solvent was removed by evaporation to give a crude product. A mixed solvent of chloroform-ethanol was added to the crude product, and recrystallization afforded 10.0 gm of the target compound as yellow crystals (yield: 64%).

Melting Point: 188.0-189.0 °C

IR( $\nu_{\text{KBr}}$ ,  $\text{cm}^{-1}$ ): 2932, 1620, 1500, 1440, 1360, 1125, 786, 640

$^1\text{H-NMR}$ ( $\text{CDCl}_3$ ,  $\delta$ ): 3.10(12H, s), 4.13(0.4H, s), 5.88(1.6H, s), 7.82(3.2H, s), 7.75(0.4H, d), 8.05(0.4H, d)

## 45 Example 13

## 50 Synthesis of 1-(4-methyl-3-oxo-4-azapentanoyl)-4-(4,4-dimethyl-3-oxopentanoyl)benzene:

In a 100 ml three-necked flask equipped with a mechanical stirrer, dropping funnel, reflux condenser, and a nitrogen-inlet tube, 5.0 gm (125 mmol) of 60% sodium hydride, 5.1 gm (51 mmol) of pinacolone, 10 gm (51 mmol) of dimethylterephthalate, and 30 ml of anhydrous tetrahydrofuran were mixed with stirring under nitrogen stream, and the refluxing under heat was continued for 4 hours. The refluxing under heat was kept for another 3 hours while adding dropwise 4.4 gm (51 mmol) of dimethyl acetamide. After cooling the reaction mixture, 65 ml of 2N aqueous hydrochloric acid was added, and the mixture was extracted with chloroform. The extract was dried over anhydrous sodium sulfate and the solvent was removed by evaporation to give a crude product. Acetone was added to the crude product, and recrystallization afforded

6.9 gm of the target compound as pale yellow crystals (yield: 43%).

Melting Point: 158.7-160.1 °C

IR( $\nu_{\text{KBr}}$ ,  $\text{cm}^{-1}$ ): 3010, 1610, 1503, 1368, 1290, 1164, 1116

$^1\text{H-NMR}(\text{CDCl}_3, \delta)$ : 1.29(9H, s), 3.10(6H, bs), 4.15(0.3H, s), 5.88(0.7H, s), 6.35(1H, s), 7.78-8.10(4H, m)

5

#### Experimental Example 1

10 UV-ray absorbing effect of the compounds of the present invention:

The absorbance as the UV-ray absorbing effect was measured on the Invention Compounds (Ia)-(Ie) prepared in Examples 1-4 and 6, Invention Compound (If)-(Ii) prepared in Examples 7-13, and Comparative Compounds shown below according to the method described below. The results are shown in Table 1.

15

#### Invention Compound:

- (Ia): 3,4-dimethoxy(4,4-dimethyl-3-oxopentanoyl)benzene
- 20 (Ib): 3,4,5-trimethoxy(4,4-dimethyl-3-oxopentanoyl)benzene
- (Ic): 1,4-bis(4,4-dimethyl-3-oxopentanoyl)benzene
- (Id): 1,3-bis(4,4-dimethyl-3-oxopentanoyl)benzene
- (Ie): 2,6-bis(4,4-dimethyl-3-oxopentanoyl) naphthalene
- (If): 1,4-bis(4-methyl-3-oxopentanoyl)benzene
- 25 (Ig): 1,4-bis(3-oxopentanoyl)benzene
- (Ih): 1,4-bis(3-oxododecanoyl)benzene
- (Ii): 1,3,5-tris(4,4-dimethyl-3-oxopentanoyl)benzene
- (Ij): 1,4-bis(4-hydroxy-4-methyl-3-oxopentanoyl)benzene
- (Ik): 1,4-bis(4-methyl-3-oxo-4-azapentanoyl)benzene
- 30 (Il): 1-(4-methyl-3-oxo-4-azapentanoyl)-4-(4,4-dimethyl-3-oxopentanoyl)benzene

#### Comparative Compound:

- 1. 2-hydroxy-4-methoxybenzophenone (commercial UV-A absorber)
- 35 2. 2-ethylhexyl p-methoxycinnamate (commercial UV-B absorber)

#### <Measuring method>

Ethanol solutions of each Invention Compound and Comparative Compound were prepared (concentration:  $2.5 \times 10^{-5}$  mol/l, ethanol: 99.5% reagent, Special grade). Each solution was injected in a quartz cell (1cm x 1cm) and its absorbance was measured using an automatic recording spectrophotometer (U-3410 model; manufactured by Hitachi, Ltd.).

45

50

55

Table 1

(UV-ray absorbing effect)														
Wavelength (nm)	Invention Compound												Comparative Compound	
	Ia	Ib	Ic	Id	Ie	If	Ig	Ih	Ii	Ij	Ik	Il	1	2
290	0.24	0.25	0.27	0.54	0.44	0.23	0.23	0.24	0.51	0.28	0.28	0.26	0.40	0.54
300	0.29	0.36	0.38	0.74	0.29	0.34	0.34	0.34	0.71	0.41	0.37	0.36	0.28	0.60
310	0.37	0.46	0.49	0.85	0.43	0.45	0.43	0.44	0.87	0.60	0.49	0.47	0.24	0.64
320	0.49	0.52	0.62	0.84	0.55	0.57	0.54	0.59	0.89	0.78	0.57	0.56	0.25	0.52
330	0.57	0.51	0.78	0.70	0.75	0.71	0.66	0.75	0.75	0.98	0.61	0.65	0.25	0.27
340	0.55	0.43	0.87	0.44	0.91	0.81	0.75	0.85	0.56	0.99	0.57	0.66	0.21	0.08
350	0.38	0.24	0.83	0.21	0.97	0.79	0.73	0.84	0.30	0.98	0.45	0.59	0.13	0.02
360	0.21	0.13	0.68	0.05	0.88	0.67	0.64	0.71	0.12	0.61	0.29	0.44	0.06	0.01
370	0.06	0.03	0.40	0.01	0.73	0.45	0.45	0.46	0.01	0.45	0.15	0.28	0.03	0
380	0.01	0.01	0.20	0	0.55	0.26	0.31	0.26	0	0.08	0.05	0.13	0.01	0
390	0	0	0.30	0	0.26	0.06	0.16	0.06	0	0.02	0.01	0.03	0	0
400	0	0	0.01	0	0.09	0.02	0.10	0.01	0	0.01	0	0.01	0	0

As clearly shown in Table 1, the compounds of the present invention possess UV-A and UV-B absorbing effects higher than conventional compounds, 2-hydroxy-4-methoxybenzophenone and 2-ethylhexyl-p-methoxycinnamate. This is the evidence that the compounds of the present invention exhibit a superior sunscreensing effect.

#### Experimental Example 2

Stability of the compound of the present invention against UV-rays:

#### <Experiment procedure>

Each of the Invention Compounds or the Comparative Compounds was dissolved in a 99.5% ethanol/distilled water (3/2) solvent to prepare 2 mmol/l concentration solutions. Onto the solution UV-ray was irradiated for 14 hours using a xenon fading tester having a similar wave length and strength as those of sun light. After the solvent was evaporated, each sample was subjected to quantitative analysis for measuring the residual rate of the composition to determine the stability against light of each composition. The results are shown in Table 2.

Table 2

(Stability against light)		
	Residual rate (%)	
	Hours after start of irradiation	
Compound	14 hours	65 hours
Invention Composition		
Ia	>99	99
Ib	>99	99
Ic	99	96
Id	99	95
Ie	99	95
If	99	94
Ig	99	94
Ih	99	95
Ii	99	96
Ij	99	95
Ik	97	92
Il	96	92
Comparative Composition		
2	42	10
3*	73	29

\* Comparative Composition 3:  
4-methoxy-4'-t-butyl-benzoylmethane

From the results in Table 2, the compounds of the present invention apparently have an outstanding stability against UV-rays in comparison with 4-methoxy-4'-t-butyl-dibenzoyl-methane and 2-ethylhexyl-p-methoxycinnamate which are known as a general UV-A absorbing agent and UV-B absorbing agent respectively.

#### Example 14

O/W cream:

The following components were formulated according to a known method to prepare O/W creams.

# EP 0 416 564 A2

5

10

15

20

25

30

## Example 15

35

W/O cream:

The following components were formulated according to a known method to prepare W/O creams.

40

45

50

55

<Formulation>		Wt%
Invention Compounds (Ia)-(II)		2.0
Stearic acid		1.0
Lipophilic-type monostearic acid glyceride		2.0
Polyoxyethylenesorbitane monostearate		1.0
Cetyl alcohol		1.0
Stearyl alcohol		1.0
Squalane		10.0
Liquid paraffin		20.0
Vaseline		5.0
Butyl parabene		0.1
Methyl parabene		0.1
Triethanol amine		1.0
Glycerol		10.0
Perfume		q.s.
Water		Balance
Total		100.0

EP 0 416 564 A2

<Formulation>	Wt%
Invention Compound (Ia)-(II)	2.0
Sorbitan sesquioleate	4.0
Aluminum stearate	0.5
Cetyl alcohol	4.0
Liquid paraffin	16.0
Squalane	10.0
Isopropylmyristate	5.0
Sodium benzoate	0.3
Glycerol	10.0
Perfume	q.s.
Water	Balance
Total	100.0

Example 16

O/W milky lotion:

The following components were formulated according to a known method to prepare O/W milky lotions.

EP 0 416 564 A2

5

10

15

20

25

30

<Formulation>		Wt%
Invention Compounds (Ia)-(II)		3.0
Stearic acid		2.0
Sorbitan monostearate		1.5
Polyoxyethylenesorbitane monostearate		1.0
Cetyl alcohol		0.4
Stearyl alcohol		0.3
Isopropylmyristate		7.0
Squalane		5.0
Liquid paraffin		5.0
Solid paraffin		2.0
Ethyl parabene		0.1
Methyl parabene		0.1
Carbopole		0.2
Potassium hydroxide		0.4
Perfume		q.s.
Water		Balance
	Total	100.0

Example 17

35

Toilet lotion:

The following components were formulated according to a known method to prepare toilet lotions.

40

45

50

55

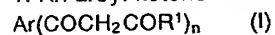


<Formulation>	Wt%
Invention Compound (Ia)-(II)	2.0
Polyoxyethylene(23 EO) laurylether	4.0
Ethanol	10.0
Glycerol	3.0
Dipropylene glycol	7.0
Lactic acid	0.05
Sodium lactate	0.12
Methyl parabene	0.1
Perfume	q.s.
Pigment	Small amount
Water	Balance
Total	100.0

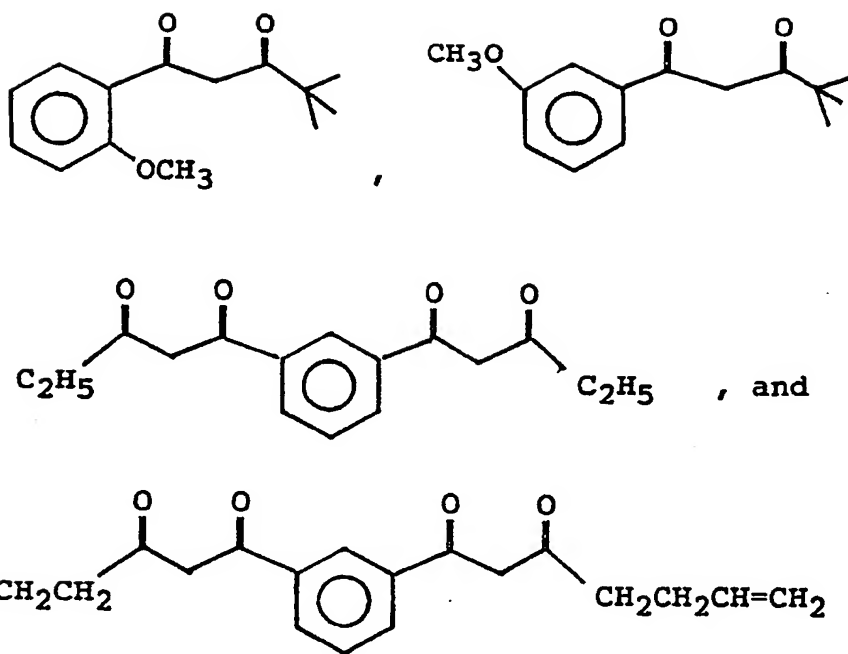
The aroyl ketone derivative (I) of the present invention can provide an excellent UV-ray absorbing effect and has an outstanding stability against light. Thus the UV-ray absorber or cosmetic composition containing the aroyl ketone derivative (I) can exhibit an excellent sunscreensing effect.

#### Claims

1. An aroyl ketone derivative represented by the following formula (I):



wherein R<sup>1</sup> represents a saturated or unsaturated hydrocarbon group having 1-24 carbon atoms, or a linear or branched alkoxyalkyl or alkenyloxyalkyl group having 1-24 carbon atoms, dialkylamino group having 1-24 carbon atoms, Ar represents a benzene or naphthalene ring which may be substituted with one or more hydroxyl groups, alkoxy groups having 1-24 carbon atoms, alkenyloxy groups having 1-24 carbon atoms, or  $\alpha$ -methylenedioxy groups, and n denotes an integer of 1-4; excluding the compounds represented by the following formulae:



2. A UV-ray absorber comprising as an effective UV-ray absorbing agent an aroyl ketone derivative represented by the following formula (I):



wherein  $\text{R}^1$  represents a saturated or unsaturated hydrocarbon group having 1-24 carbon atoms, or a linear or branched alkoxyalkyl or alkenyloxyalkyl group having 1-24 carbon atoms, dialkylamino group having 1-24 carbon atoms, Ar represents a benzene or naphthalene ring which may be substituted with one or more hydroxyl groups, alkoxy groups having 1-24 carbon atoms, alkenyloxy groups having 1-24 carbon atoms, polyoxylenedioxy groups, or  $\alpha$ -methylenedioxy groups, and n denotes an integer of 1-4.

3. A cosmetic composition comprising (a) an aroyl ketone derivative represented by the following formula (I):



wherein  $\text{R}^1$  represents a saturated or unsaturated hydrocarbon group having 1-24 carbon atoms, or a linear or branched alkoxyalkyl or alkenyloxyalkyl group having 1-24 carbon atoms, dialkylamino group having 1-24 carbon atoms, Ar represents a benzene or naphthalene ring which may be substituted with one or more hydroxyl groups, alkoxy groups having 1-24 carbon atoms, alkenyloxy groups having 1-24 carbon atoms, polyoxylenedioxy groups, or  $\alpha$ -methylenedioxy groups, and n denotes an integer of 1-4; and (b) cosmetically acceptable components.

(19)



Europäisches Patentamt

European Patent Office

Office européen des brevets



(11) Publication number:

**0 416 564 A3**

(12)

**EUROPEAN PATENT APPLICATION**(21) Application number: **90117020.9**(22) Date of filing: **04.09.90**

(51) Int. Cl.<sup>5</sup>: **C07C 49/835**, C07C 49/84,  
C07C 49/76, C07C 49/788,  
C07C 49/825, C07C 49/83,  
C07C 49/796, C07C 49/82,  
C07C 49/794, C07C 235/78,  
A61K 7/42

(30) Priority: **05.09.89 JP 229708/89**  
**22.01.90 JP 12196/90**

(43) Date of publication of application:  
**13.03.91 Bulletin 91/11**

(64) Designated Contracting States:  
**DE ES FR GB**

(88) Date of deferred publication of the search report:  
**11.03.92 Bulletin 92/11**

(71) Applicant: **KAO CORPORATION**  
**1-14-10, Nihonbashikayaba-cho**  
**Chuo-ku Tokyo(JP)**

(72) Inventor: **Yamada, Shinji**  
**4594, Ichihana, Ichikaimachi**  
**Haga-gun, Tochigi(JP)**  
Inventor: **Kawamata, Akira**  
**8-3, 1-chome, Daishinden**

**Utsunomiya-shi, Tochigi(JP)**Inventor: **Imokawa, Genji****1022-89, Himuro-cho****Utsunomiya-shi, Tochigi(JP)**Inventor: **Masuda, Shinichi****6-100, Dojojuku-machi****Utsunomiya-shi, Tochigi(JP)**Inventor: **Yamaguchi, Masakazu****4594, Ichihana, Ichikai-machi****Haga-gun, Tochigi(JP)**Inventor: **Niinaka, Koichi****4594, Ichihana, Ichikai-machi****Haga-gun, Tochigi(JP)**Inventor: **Joukura, Hiroko****561-10, Yamamoto-cho****Utsunomiya-shi, Tochigi(JP)**

(74) Representative: **Wächtershäuser, Günter, Dr.**  
**Tal 29**  
**W-8000 München 2(DE)**

(54) **Novel aroyl ketone derivative, UV ray absorber comprising the same, and cosmetic composition containing the same.**

(57) A novel aroyl ketone derivative represented by formula (I) is disclosed.

$\text{Ar}(\text{COCH}_2\text{COR}^1)_n$  (I)

wherein R<sup>1</sup> represents a saturated or unsaturated hydrocarbon group having 1-24 carbon atoms, or a linear or branched alkoxyalkyl or alkenyloxyalkyl group having 1-24 carbon atoms, dialkylamino group having 1-24 carbon atoms, Ar represents a benzene or naphthalene ring which may be substituted with one or more hydroxyl groups, alkoxy groups having 1-24 carbon atoms, alkenyloxy groups having 1-24 carbon atoms, polyoxyleneoxy groups, or  $\alpha$ -methylenedioxy groups, and n denotes an integer of

1-4. An UV-ray absorber and cosmetic composition comprising the aroyl ketone derivative of formula (I) are also disclosed. Aroyl ketone derivative (I) of the present invention can provide an excellent UV-ray absorbing effect and has outstanding stability against light, and the UV-ray absorber or cosmetic composition containing the aroyl ketone derivative (I) can exhibit an excellent sunscreensing effect.

**EP 0 416 564 A3**





European  
Patent Office

## EUROPEAN SEARCH REPORT

Application Number

EP 90 11 7020

### DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.5)
X	EP-A-0 319 325 (WAKO PURE CHEMICAL INDUSTRIES) * page 9, compound VIII; page 8, lines 10-11 * - - -	1	C 07 C 49/835 C 07 C 49/84 C 07 C 49/76
X	EP-A-0 323 050 (WAKO PURE CHEMICAL INDUSTRIES) * page 14, compound II-1; page 72, line 35 * - - -	1	C 07 C 49/788 C 07 C 49/825 C 07 C 49/83
A,D	JOURNAL OF THE AMERICAN CHEMICAL SOCIETY vol. 80, no. 18, 1958, pages 4891-4895, US; D.F. MARTIN et al.: "Bis-(beta-diketones).I. Synthesis of compounds of the type RCOCH <sub>2</sub> CO-Y-COCH <sub>2</sub> COR <sub>1,2</sub> " * pages 4891-4892 * - - - - -	1	C 07 C 49/796 C 07 C 49/82 C 07 C 49/794 C 07 C 235/78 A 61 K 7/42 C 07 C 233/11 A 61 K 7/00
The present search report has been drawn up for all claims			TECHNICAL FIELDS SEARCHED (Int. Cl.5)
			A 61 K 7/42 C 07 C 49/76 C 07 C 49/788 C 07 C 49/794 C 07 C 49/796 C 07 C 49/82 C 07 C 49/825 C 07 C 49/83 C 07 C 49/835 C 07 C 49/84 C 07 C 235/78
Place of search		Date of completion of search	Examiner
Berlin		27 November 91	KNAACK
<b>CATEGORY OF CITED DOCUMENTS</b>			
X: particularly relevant if taken alone			
Y: particularly relevant if combined with another document of the same category			
A: technological background			
O: non-written disclosure			
P: intermediate document			
T: theory or principle underlying the invention			
E: earlier patent document, but published on, or after the filing date			
D: document cited in the application			
L: document cited for other reasons			
&: member of the same patent family, corresponding document			

